

## **GHS Safety Data Sheet**

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## Section 1 - CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

## PRODUCT NAME

ISOBUTYL IODIDE

#### OTHER NAMES

C4-H9-I, CH3CH2(CH3)CH2I, "propane, 1-iodo-2-methyl-", "isobutyl iodide", "primary isobutyl iodide"

#### PROPER SHIPPING NAME

**IODOMETHYLPROPANES** 

## **PRODUCT USE**

■ Intermediate.

#### **SUPPLIER**

Company: S D FINE- CHEM LIMITED

Address:

315- 317, T.V.Ind.Estate, 248. Worli Road. Mumbai- 400030, India www.sdfine.com

Telephone: 91-22 24959898/99 Fax: 91- 22 2493 7232 Email: technical@sdfine.com

# Section 2 - HAZARDS IDENTIFICATION

# **GHS Classification**

Acute Toxicity Category 4 Eye Irritation Category 2A Flammable Liquid Category 2 Skin Corrosion/Irritation Category 2 STOT - SE Category 3





# **EMERGENCY OVERVIEW**

### **HAZARD**

**DANGER** 

# Determined using GHS criteria

H225 Highly flammable liquid and vapour.

H315 Causes skin irritation.

H319 Causes serious eye irritation.

H332 Harmful if inhaled.

H335 May cause respiratory irritation.

# PRECAUTIONARY STATEMENTS

Prevention

Code

P210 Keep away from heat/sparks/open flames/hot surfaces. - No smoking.

P233 Keep container tightly closed.

P240 Ground/bond container and receiving equipment.

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P241 Use explosion- proof electrical/ventilating/lighting/ ... /equipment

P242 Use only non- sparking tools.

P243 Take precautionary measures against static discharge.
P261 Avoid breathing dust/ fume/ gas/ mist/ vapours/ spray.

P264 Wash ... thoroughly after handling.

P271 Use only outdoors or in a well- ventilated area.

P280 Wear protective gloves/protective clothing/eye protection/face protection.

Response

Code Phrase

P302+P352 IF ON SKIN: Wash with plenty of soap and water.

P303+P361+P353 IF ON SKIN (or hair): Remove/Take off immediately all contaminated

clothing. Rinse skin with water/shower.

P304+P340 IF INHALED: Remove victim to fresh air and keep at rest in a position

comfortable for breathing.

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove

contact lenses, if present and easy to do. Continue rinsing.

P312 Call a POISON CENTER or doctor/physician if you feel unwell.

P332+P313 If skin irritation occurs: Get medical advice/attention.
P337+P313 If eye irritation persists: Get medical advice/attention.
P362 Take off contaminated clothing and wash before re- use.

Storage

Code Phrase

P403+P233 Store in a well- ventilated place. Keep container tightly closed.

P403+P235 Store in a well- ventilated place. Keep cool.

P405 Store locked up.

Disposal

Code Phrase

P501 Dispose of contents/container to ...

## Section 3 - COMPOSITION / INFORMATION ON INGREDIENTS

NAME
CAS RN
1- iodo- 2- methylpropane
stabilised with (typically) copper

CAS RN
513-38-2
>98

# Section 4 - FIRST AID MEASURES

# **SWALLOWED**

- If poisoning occurs, contact a doctor or Poisons Information Centre.
- · Avoid giving milk or oils.
- · Avoid giving alcohol.
- If swallowed do NOT induce vomiting.
- If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.
- · Observe the patient carefully.
- Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.

# EYE

- If this product comes in contact with the eyes:
- · Wash out immediately with fresh running water.
- Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.
- Seek medical attention without delay; if pain persists or recurs seek medical attention.
- Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.

#### SKIN

- If skin contact occurs:
- Immediately remove all contaminated clothing, including footwear.
- · Flush skin and hair with running water (and soap if available).
- · Seek medical attention in event of irritation.

### **INHALED**

- · If fumes or combustion products are inhaled remove from contaminated area.
- · Lay patient down. Keep warm and rested.
- Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid

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procedures.

 Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.

# **NOTES TO PHYSICIAN**

■ Treat symptomatically.

for intoxication due to Freons/ Halons;

A: Emergency and Supportive Measures

- · Maintain an open airway and assist ventilation if necessary
- Treat coma and arrhythmias if they occur. Avoid (adrenaline) epinephrine or other sympathomimetic amines that may precipitate
  ventricular arrhythmias. Tachyarrhythmias caused by increased myocardial sensitisation may be treated with propranolol, 1-2 mg
  IV or esmolol 25-100 microgm/kg/min IV.
- · Monitor the ECG for 4-6 hours

B: Specific drugs and antidotes:

There is no specific antidote.

DO NOT administer sympathomimetic drugs as they may cause ventricular arrhythmias.

### **Section 5 - FIRE FIGHTING MEASURES**

#### **EXTINGUISHING MEDIA**

- · Water spray or fog.
- · Foam.
- · Dry chemical powder.
- · BCF (where regulations permit).

#### **FIRE FIGHTING**

- · Alert Fire Brigade and tell them location and nature of hazard.
- · May be violently or explosively reactive.
- · Wear breathing apparatus plus protective gloves in the event of a fire.
- Prevent, by any means available, spillage from entering drains or water course.

# FIRE/EXPLOSION HAZARD

- · Liquid and vapour are highly flammable.
- Severe fire hazard when exposed to heat, flame and/or oxidisers.
- Vapour may travel a considerable distance to source of ignition.
- Heating may cause expansion or decomposition leading to violent rupture of containers.

Combustion products include: carbon dioxide (CO2), hydrogen iodide, other pyrolysis products typical of burning organic material.

# FIRE INCOMPATIBILITY

Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may
result.

### **Section 6 - ACCIDENTAL RELEASE MEASURES**

### MINOR SPILLS

- · Remove all ignition sources.
- · Clean up all spills immediately.
- Avoid breathing vapours and contact with skin and eyes.
- · Control personal contact with the substance, by using protective equipment.

### **MAJOR SPILLS**

- · Clear area of personnel and move upwind.
- · Alert Fire Brigade and tell them location and nature of hazard.
- May be violently or explosively reactive.
- Wear breathing apparatus plus protective gloves.

Personal Protective Equipment advice is contained in Section 8 of the MSDS.

# **Section 7 - HANDLING AND STORAGE**

# PROCEDURE FOR HANDLING

- · Containers, even those that have been emptied, may contain explosive vapours.
- Do NOT cut, drill, grind, weld or perform similar operations on or near containers.
- DO NOT allow clothing wet with material to stay in contact with skin.

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- · Avoid all personal contact, including inhalation.
- · Wear protective clothing when risk of exposure occurs.
- · Use in a well-ventilated area.
- · Prevent concentration in hollows and sumps.

#### SUITABLE CONTAINER

- DO NOT use aluminium or galvanised containers.
- · Packing as supplied by manufacturer.
- Plastic containers may only be used if approved for flammable liquid.
- · Check that containers are clearly labelled and free from leaks.
- For low viscosity materials (i): Drums and jerry cans must be of the non-removable head type. (ii): Where a can is to be used as an inner package, the can must have a screwed enclosure.
- For materials with a viscosity of at least 2680 cSt. (23 deg. C)
- For manufactured product having a viscosity of at least 250 cSt. (23 deg. C)
- Manufactured product that requires stirring before use and having a viscosity of at least 20 cSt (25 deg. C): (i) Removable head packaging; (ii) Cans with friction closures and (iii) low pressure tubes and cartridges may be used.

#### STORAGE INCOMPATIBILITY

- Haloalkanes:
- are highly reactive:some of the more lightly substituted lower members are highly flammable; the more highly substituted may be used as fire suppressants, not always with the anticipated results.
- may react with the lighter divalent metals to produce more reactive compounds analogous to Grignard reagents.
- · may produce explosive compounds following prolonged contact with metallic or other azides
- may react on contact with potassium or its alloys although apparently stable on contact with a wide rage of halocarbons, reaction products may be shock-sensitive and may explode with great violence on light impact; severity generally increases with the degree of halocarbon substitution and potassium-sodium alloys give extremely sensitive mixtures.
- · Avoid strong bases.

#### STORAGE REQUIREMENTS

- · Store in original containers in approved flame-proof area.
- · No smoking, naked lights, heat or ignition sources.
- DO NOT store in pits, depressions, basements or areas where vapours may be trapped.
- Keep containers securely sealed.

May discolour on exposure to light.

## Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

#### **EXPOSURE CONTROLS**

The following materials had no OELs on our records

• 1- iodo- 2- methylpropane:

CAS:513-38-2

### **MATERIAL DATA**

1-IODO-2-METHYLPROPANE:

■ Sensory irritants are chemicals that produce temporary and undesirable side-effects on the eyes, nose or throat. Historically occupational exposure standards for these irritants have been based on observation of workers' responses to various airborne concentrations.

# PERSONAL PROTECTION









#### EYE

- · Safety glasses with side shields.
- · Chemical goggles.
- Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lens or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent].

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# Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

#### HANDS/FEET

- · Wear chemical protective gloves, e.g. PVC.
- · Wear safety footwear or safety gumboots, e.g. Rubber.

The selection of the suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application.

The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice.

Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include:.

Neoprene gloves.

#### **OTHER**

- · Overalls.
- PVC Apron.
- PVC protective suit may be required if exposure severe.
- Eyewash unit.
- Some plastic personal protective equipment (PPE) (e.g. gloves, aprons, overshoes) are not recommended as they may produce static electricity.
- For large scale or continuous use wear tight-weave non-static clothing (no metallic fasteners, cuffs or pockets), non sparking safety footwear.

#### **ENGINEERING CONTROLS**

■ Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment.

# **Section 9 - PHYSICAL AND CHEMICAL PROPERTIES**

## **APPEARANCE**

Colourless liquid; does not mix with water. Soluble in alcohol, chloroform, ether. May discolour on standing.

# **PHYSICAL PROPERTIES**

Liquid.

Does not mix with water.

Sinks in water.

State	Liquid	Molecular Weight	184.02
Melting Range (°C)	- 9 <sup>3</sup>	Viscosity	Not Available
Boiling Range (°C)	120- 121	Solubility in water (g/L)	Immiscible
Flash Point (°C)	12.78	pH (1% solution)	Not applicable
Decomposition Temp (°C)	Not Available	pH (as supplied)	Not applicable
Autoignition Temp (°C)	Not available	Vapour Pressure (kPa)	2.66 @ 20 C
Upper Explosive Limit (%)	Not available	Specific Gravity (water=1)	1.599
Lower Explosive Limit (%)	Not available	Relative Vapour Density	6

(air=1)

Volatile Component (%vol) Not available Evaporation Rate Not available

# Section 10 - CHEMICAL STABILITY AND REACTIVITY INFORMATION

# **CONDITIONS CONTRIBUTING TO INSTABILITY**

- · Presence of incompatible materials.
- · Product is considered stable.
- · Hazardous polymerisation will not occur.

For incompatible materials - refer to Section 7 - Handling and Storage.

# **Section 11 - TOXICOLOGICAL INFORMATION**

#### Health hazard summary table:

Acute toxicity
Skin corrosion/irritation
Serious eye damage/irritation
Respiratory or skin sensitization
Germ cell mutagenicity
Carcinogenicity
Reproductive toxicity
STOT- single exposure
STOT- repeated exposure
Aspiration hazard

Acute Tox. (inhal) 4 Skin Irrit. 2 Eye Irrit. 2A Not applicable Not applicable Not applicable Not applicable STOT SE 3 Not applicable Not applicable

### POTENTIAL HEALTH EFFECTS

## **ACUTE HEALTH EFFECTS**

# SWALLOWED

■ The material is not thought to produce adverse health effects following ingestion (as classified by EC Directives using animal models).

Nevertheless, adverse systemic effects have been produced following exposure of animals by at least one other route and good hygiene practice requires that exposure be kept to a minimum.

#### **EYE**

■ This material can cause eye irritation and damage in some persons.

#### SKIN

- This material can cause inflammation of the skin on contact in some persons.
- The material may accentuate any pre-existing dermatitis condition.
- Repeated exposure may cause skin cracking, flaking or drying following normal handling and use.
- Skin contact with the material may damage the health of the individual; systemic effects may result following absorption.
- Open cuts, abraded or irritated skin should not be exposed to this material.
- Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.

### **INHALED**

- Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may be harmful
- The material can cause respiratory irritation in some persons.

The body's response to such irritation can cause further lung damage.

- Inhalation of vapours may cause drowsiness and dizziness.
- This may be accompanied by sleepiness, reduced alertness, loss of reflexes, lack of co-ordination, and vertigo.
- Inhalation hazard is increased at higher temperatures.
- Acute intoxication by halogenated aliphatic hydrocarbons appears to take place over two stages.

Signs of a reversible narcosis are evident in the first stage and in the second stage signs of injury to organs may become evident, a single organ alone is (almost) never involved.

- Depression of the central nervous system is the most outstanding effect of most halogenated aliphatic hydrocarbons. Inebriation and excitation, passing into narcosis, is a typical reaction.
- In the case of iodised and brominated compounds, exposure effects cannot be described by simple central nervous system depression produced by other halogenated aliphatic hydrocarbons.

Headache, nausea, ataxia (loss of muscle co-ordination), tremors, speech difficulties, visual disturbances, convulsions, paralysis, delirium, mania and apathy are all evidence of additional effects.

### **CHRONIC HEALTH EFFECTS**

■ Long-term exposure to respiratory irritants may result in disease of the airways involving difficult breathing and related systemic problems.

Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure.

Chronic solvent inhalation exposures may result in nervous system impairment and liver and blood changes. [PATTYS].

### **TOXICITY AND IRRITATION**

■ Asthma-like symptoms may continue for months or even years after exposure to the material ceases. This may be due to a non-allergenic condition known as reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound.

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# **Section 12 - ECOLOGICAL INFORMATION**

This material and its container must be disposed of as hazardous waste.

**Ecotoxicity** 

Ingredient Persistence: Persistence: Air Bioaccumulation Mobility Water/Soil

1- iodo- 2- methylpropane HIGH No Data LOW HIGH

Available

# **Section 13 - DISPOSAL CONSIDERATIONS**

■ Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area.

A Hierarchy of Controls seems to be common - the user should investigate:

- · Reduction.
- DO NOT allow wash water from cleaning or process equipment to enter drains.
- It may be necessary to collect all wash water for treatment before disposal.
- In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.
- Where in doubt contact the responsible authority.
- · Recycle wherever possible.
- Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.
- Dispose of by: burial in a land-fill specifically licenced to accept chemical and / or pharmaceutical wastes or Incineration in a licenced apparatus (after admixture with suitable combustible material).
- Decontaminate empty containers. Observe all label safeguards until containers are cleaned and destroyed.

# **Section 14 - TRANSPORTATION INFORMATION**



Labels Required: FLAMMABLE LIQUID

HAZCHEM:

2YE

Land Transport UNDG:

Class or division: 3 Subsidiary risk: None UN No.: 2391 UN packing group: II

Shipping Name:IODOMETHYLPROPANES

Air Transport IATA:

ICAO/IATÀ Class: 3 ICAO/IATA Subrisk: None UN/ID Number: 2391 Packing Group: II Special provisions: None

Shipping name:IODOMETHYLPROPANES

**Maritime Transport IMDG:** 

IMDG Class:3IMDG Subrisk:NoneUN Number:2391Packing Group:IIEMS Number:F- E, S- DSpecial provisions:None

Limited Quantities: 1 L
Shipping name:IODOMETHYLPROPANES

## **Section 15 - REGULATORY INFORMATION**

## **REGULATIONS**

# 1-iodo-2-methylpropane (CAS: 513-38-2) is found on the following regulatory lists;

"FisherTransport Information","India Chemical Accidents Rules - Schedule 2: Threshold Quantities","India Chemical Accidents Rules - Schedule 3: Named Chemicals","India Hazardous Wastes (Management, Handling and Transboundary Movement) Rules - Schedule 2: List of Wastes Constituents with Concentration Limits","India Manufacture, Storage and Import of Hazardous Chemical Rules - Schedule 2: Isolated storage at Installations other than those covered by Schedule 4","India Manufacture, Storage and Import of Hazardous Chemical Rules - Schedule 3: List of Hazardous Chemicals for Application of Rules 5 and 7 to 15","Sigma-AldrichTransport Information"

# **Section 16 - OTHER INFORMATION**

# Denmark Advisory list for selfclassification of dangerous substances

Substance CAS Suggested codes 1- iodo- 2- methylpropane 513- 38- 2 Xn; R22

- Classification of the preparation and its individual components has drawn on official and authoritative sources using available literature references.
- The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings.

The above information is believed to be accurate and represent the best information currently available to us, but does not represent any warranty expressed or implied of the properties of the product. User should make their own investigation to determine the suitability of the information for their particular purpose.

Issue Date: 8-Jun-2018