

GHS Safety Data Sheet

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### Section 1 - CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

### **PRODUCT NAME**

ANTIMONY SULPHIDE BALCK

#### **OTHER NAMES**

Sb2-S3, "antimonous sulfide", "antimony glance", "antimony orange", "antimony sesquisulfide", "antimony trisulphide, colloid", "antimony vermilion", "black antimony", "C.I. 77060", "C.I. Pigment Red 107", "crimson antimony", "diantimony trisulfide", lymphoscan, "needle antimony"

#### PROPER SHIPPING NAME

ANTIMONY COMPOUND, INORGANIC, SOLID, N.O.S.(contains antimony trisulfide)

#### **PRODUCT USE**

Used in pyrotechnics; Bengal fires; in the manufacture of ruby glass, matches, percussion caps, explosives; as a pigment in paints especially in camouflage paints where it reflects infrared radiation in the same way as green vegetation. Food

#### **SUPPLIER**

Company: S D FINE- CHEM LIMITED Address: 315- 317, T.V.Ind.Estate, 248, Worli Road, Mumbai- 400030, India www.sdfine.com Telephone: 91- 22 24959898/99

Telephone: 91- 22 24959898/99 Fax: 91- 22 2493 7232 Email: technical@sdfine.com

# **Section 2 - HAZARDS IDENTIFICATION**

### **GHS Classification**

Acute Toxicity Category 4
Acute Toxicity Category 4
Chronic Aquatic Hazard Category 1





#### **EMERGENCY OVERVIEW**

### **HAZARD**

WARNING

Determined by using GHS criteria

H302 Harmful if swallowed. H332 Harmful if inhaled.

H410 Very toxic to aquatic life with long lasting effects.

### PRECAUTIONARY STATEMENTS

Prevention

Code Phrase

P261 Avoid breathing dust/ fume/ gas/ mist/ vapours/ spray.

P264 Wash ... thoroughly after handling.

P270 Do not eat, drink or smoke when using this product.

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**Section 2 - HAZARDS IDENTIFICATION** 

P271 Use only outdoors or in a well- ventilated area.

P273 Avoid release to the environment.

Response

P304+P340

Code Phrase

P301+P312 IF SWALLOWED: Call a POISON CENTER or doctor/physician if you feel unwell.

IF INHALED: Remove victim to fresh air and keep at rest in a position

comfortable for breathing.

Call a POISON CENTER or doctor/physician if you feel unwell.

P330 Rinse mouth. P391 Collect spillage.

**Disposal** 

P312

Code Phrase

P501 Dispose of contents/container to ...

### Section 3 - COMPOSITION / INFORMATION ON INGREDIENTS

NAME CAS RN % antimony trisulfide 1345-04-6 >95

# **Section 4 - FIRST AID MEASURES**

#### **SWALLOWED**

- IF SWALLOWED, REFER FOR MEDICAL ATTENTION, WHERE POSSIBLE, WITHOUT DELAY.
- For advice, contact a Poisons Information Centre or a doctor.
- · Urgent hospital treatment is likely to be needed.
- In the mean time, qualified first-aid personnel should treat the patient following observation and employing supportive measures as indicated by the patient's condition.

#### EYE

- If this product comes in contact with the eyes:
- Wash out immediately with fresh running water.
- Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.
- Seek medical attention without delay; if pain persists or recurs seek medical attention.
- Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.

#### SKIN

- If skin contact occurs:
- Immediately remove all contaminated clothing, including footwear.
- Flush skin and hair with running water (and soap if available).
- Seek medical attention in event of irritation.

## INHALED

- If fumes or combustion products are inhaled remove from contaminated area.
- Lay patient down. Keep warm and rested.
- Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.
- Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.

### **NOTES TO PHYSICIAN**

- For exposures involving sulfides and hydrogen sulfide (including gastric acid decomposition products of alkaline sulfides):
- Hydrogen sulfide anion produces its major toxic effect through inhibition of cytochrome oxidases.
- Symptoms include profuse salivation, nausea, vomiting and diarrhea. Central nervous effects may include giddiness, headache, vertigo, amnesia, confusion and unconsciousness. Tachypnoea, palpitations, tachycardia, arrhythmia, sweating, weakness and muscle cramps may also indicate overexposure.

Treatment involves:

- If respirations are depressed, application of artificial respiration, administration of oxygen (continue after spontaneous breathing is established).
- For severe poisonings administer amyl nitrite and sodium nitrite (as for cyanide poisoning) but omit sodium thiosulfate injection.
- · Chelation with British Anti-Lewisite (BAL) for serious antimony exposures should be employed.
- Dialyse as needed. The role of exchange diffusion is not clear.
- Be sure to monitor for dysrhythmias.

[Ellenhorn and Barceloux: Medical Toxicology].

Copper, magnesium, aluminium, antimony, iron, manganese, nickel, zinc (and their compounds) in welding, brazing, galvanising or

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Section 4 - FIRST AID MEASURES

smelting operations all give rise to thermally produced particulates of smaller dimension than may be produced if the metals are divided mechanically. Where insufficient ventilation or respiratory protection is available these particulates may produce "metal fume fever" in workers from an acute or long term exposure.

- Onset occurs in 4-6 hours generally on the evening following exposure. Tolerance develops in workers but may be lost over the weekend. (Monday Morning Fever)
- Pulmonary function tests may indicate reduced lung volumes, small airway obstruction and decreased carbon monoxide diffusing capacity but these abnormalities resolve after several months.

### **Section 5 - FIRE FIGHTING MEASURES**

#### **EXTINGUISHING MEDIA**

- · Water spray or fog.
- Foam.
- · Dry chemical powder.
- · BCF (where regulations permit).

#### **FIRE FIGHTING**

- Alert Fire Brigade and tell them location and nature of hazard.
- Wear full body protective clothing with breathing apparatus.
- Prevent, by any means available, spillage from entering drains or water course.
- Use fire fighting procedures suitable for surrounding area.

## FIRE/EXPLOSION HAZARD

- · Non combustible.
- Not considered a significant fire risk, however containers may burn.

Decomposition may produce toxic fumes of: sulfur oxides (SOx), hydrogen sulfide (H2S), metal oxides. May emit poisonous fumes.

### FIRE INCOMPATIBILITY

■ None known.

#### Section 6 - ACCIDENTAL RELEASE MEASURES

### **MINOR SPILLS**

- · Remove all ignition sources.
- · Clean up all spills immediately.
- Avoid contact with skin and eyes.
- Control personal contact with the substance, by using protective equipment.

#### **MAJOR SPILLS**

- Clear area of personnel and move upwind.
- · Alert Fire Brigade and tell them location and nature of hazard.
- Wear full body protective clothing with breathing apparatus.
- Prevent, by any means available, spillage from entering drains or water course.

Personal Protective Equipment advice is contained in Section 8 of the MSDS.

# Section 7 - HANDLING AND STORAGE

# PROCEDURE FOR HANDLING

- Avoid all personal contact, including inhalation.
- Wear protective clothing when risk of exposure occurs.
- Use in a well-ventilated area.
- · Prevent concentration in hollows and sumps.

# **SUITABLE CONTAINER**

- · Lined metal can, lined metal pail/ can.
- · Plastic pail.
- · Polyliner drum.
- Packing as recommended by manufacturer.

For low viscosity materials

- Drums and jerricans must be of the non-removable head type.
- Where a can is to be used as an inner package, the can must have a screwed enclosure. <</>

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### **Section 7 - HANDLING AND STORAGE**

#### STORAGE INCOMPATIBILITY

- Contact with acids produces toxic fumes.
- Sulfides are incompatible with acids, diazo and azo compounds, halocarbons, isocyanates, aldehydes, alkali metals, nitrides, hydrides, and other strong reducing agents.
- Many reactions of sulfides with these materials generate heat and in many cases hydrogen gas.
- · Many sulfide compounds may liberate hydrogen sulfide upon reaction with an acid.
- · Metals and their oxides or salts may react violently with chlorine trifluoride and bromine trifluoride.
- These trifluorides are hypergolic oxidisers. They ignites on contact (without external source of heat or ignition) with recognised fuels - contact with these materials, following an ambient or slightly elevated temperature, is often violent and may produce ignition.
- The state of subdivision may affect the results.

# STORAGE REQUIREMENTS

- · Store in original containers.
- · Keep containers securely sealed.
- · Store in a cool, dry, well-ventilated area.
- Store away from incompatible materials and foodstuff containers.

### Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

### **EXPOSURE CONTROLS**

The following materials had no OELs on our records

antimony trisulfide:

CAS:1345-04-6

#### **MATERIAL DATA**

ANTIMONY TRISULFIDE:

■ The wide-ranging effects of antimony compounds have made it difficult to recommend an exposure standard which characterises the toxicology of these substances. One criteria, reflecting the irritant properties of antimony pentachloride, produced a calculated value of 5.0 mg/m3 (as antimony), which on the basis of experience was felt to be too high but did act as an "out-rider".

### PERSONAL PROTECTION









# RESPIRATOR

•Particulate. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

#### **EYE**

- · Safety glasses with side shields.
- · Chemical goggles.
- Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lens or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent].

#### HANDS/FEET

- Wear chemical protective gloves, e.g. PVC.
- Wear safety footwear or safety gumboots, e.g. Rubber.

The selection of the suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application.

The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice.

Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include:

### **OTHER**

- Overalls.
- Eyewash unit.
- Barrier cream.

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#### Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

• Skin cleansing cream.

### **ENGINEERING CONTROLS**

■ Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment.

### Section 9 - PHYSICAL AND CHEMICAL PROPERTIES

#### **APPEARANCE**

Occurs in two forms.

(a): Black crystalline powder. (b): Orange-red crystals.

Does not mix with water. Soluble in concentrated hydrochloric acid with evolution of H2S. Soluble in solutions of fixed alkali

hvdroxides. No odour. Technical or natural grades are black. Not Dangerous Goods if contains <0.5% arsenic.

WARNING: Explosion risk when in contact with oxidizing materials.

### **PHYSICAL PROPERTIES**

Solid.

Does not mix with water.

Sinks in water.

State	Divided solid	Molecular Weight	339.72
Melting Range (°C)	550 (decomp)	Viscosity	Not Applicable
Boiling Range (°C)	1150	Solubility in water (g/L)	Immiscible
Flash Point (°C)	Not Applicable	pH (1% solution)	Not available
Decomposition Temp (°C)	Not Available	pH (as supplied)	Not applicable
Autoignition Temp (°C)	Not Applicable	Vapour Pressure (kPa)	Negligible
Upper Explosive Limit (%)	Not Applicable	Specific Gravity (water=1)	4.6 approx
Lower Explosive Limit (%)	Not Applicable	Relative Vapour Density	Not applicable

(air=1)

Volatile Component (%vol) Negligible Evaporation Rate Not applicable

# Section 10 - CHEMICAL STABILITY AND REACTIVITY INFORMATION

# CONDITIONS CONTRIBUTING TO INSTABILITY

- Presence of incompatible materials.
- Product is considered stable.
- Hazardous polymerisation will not occur.

For incompatible materials - refer to Section 7 - Handling and Storage.

## **Section 11 - TOXICOLOGICAL INFORMATION**

# Health hazard summary table:

Acute toxicity Acute Tox. (inhal) 4 Acute Tox. (oral) 4 Skin corrosion/irritation Not applicable Serious eye damage/irritation Not applicable Respiratory or skin sensitization Not applicable Germ cell mutagenicity Not applicable Carcinogenicity Not applicable Reproductive toxicity Not applicable STOT- single exposure Not applicable STOT- repeated exposure Not applicable Not applicable Aspiration hazard

### **POTENTIAL HEALTH EFFECTS**

#### **ACUTE HEALTH EFFECTS**

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### **Section 11 - TOXICOLOGICAL INFORMATION**

#### **SWALLOWED**

- Accidental ingestion of the material may be harmful; animal experiments indicate that ingestion of less than 150 gram may be fatal or may produce serious damage to the health of the individual.
- Antimony poisoning causes similar symptoms to arsenic poisoning although vomiting is more prominent.

There may be changes in the rhythm of the heart beat.

■ If ingested, sulfide salts can form hydrogen sulfide, causing headache, cyanosis, low blood pressure, loss of consciousness, tremors and convulsions.

#### **EYE**

■ Although the material is not thought to be an irritant (as classified by EC Directives), direct contact with the eye may cause transient discomfort characterised by tearing or conjunctival redness (as with windburn). Slight abrasive damage may also result.

#### SKIN

- There is some evidence to suggest that this material can cause inflammation of the skin on contact in some persons.
- Skin contact with antimony compounds may result in redness and severe irritation with the formation of itchy papules, pustules, skin lesions/ small septic blisters (antimony spots) within a few hours.

  Rhinitis may also result from dermal contact.
- Open cuts, abraded or irritated skin should not be exposed to this material.
- Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.

#### **INHALED**

- Inhalation of dusts, generated by the material, during the course of normal handling, may be harmful.
- The material is not thought to produce respiratory irritation (as classified by EC Directives using animal models). Nevertheless inhalation of dusts, or fumes, especially for prolonged periods, may produce respiratory discomfort and occasionally, distress.
- Persons with impaired respiratory function, airway diseases and conditions such as emphysema or chronic bronchitis, may incur further disability if excessive concentrations of particulate are inhaled.
- If prior damage to the circulatory or nervous systems has occurred or if kidney damage has been sustained, proper screenings should be conducted on individuals who may be exposed to further risk if handling and use of the material result□□in excessive exposures.
- The inhalation of small particles of metal oxide results in sudden thirst, a sweet, metallic foul taste, throat irritation, cough, dry mucous membranes, tiredness and general unwellness.
- Headache, nausea and vomiting, fever or chills, restlessness, sweating, diarrhoea, excessive urination and prostration may also occur.
- Inhalation of antimony can cause breathing difficulties and gastrointestinal upset including sore throat, shallow breathing, dizziness, weight loss, gum bleeds and anaemia. Lung swelling and congestion can occur.

### **CHRONIC HEALTH EFFECTS**

■ There has been some concern that this material can cause cancer or mutations but there is not enough data to make an assessment. Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure.

Long term exposure to high dust concentrations may cause changes in lung function i.e. pneumoconiosis; caused by particles less than 0.5 micron penetrating and remaining in the lung. Prime symptom is breathlessness; lung shadows show on X-ray. Repeated or prolonged exposure to antimony and its compounds may produce inflammation of the mouth cavity, dry throat, metallic taste, gum infection, perforation of the nasal septum and throat, laryngitis, headache, difficulty breathing, indigestion, nausea, vomiting, diarrhoea, loss of appetite, anaemia, weight loss, tightness and pain in the chest, sleeplessness, muscular pain and weakness, dizziness, pharyngitis, bronchitis and pneumonia. Degenerative changes of the liver and kidney may occur. Chronic exposure to antimony compounds may result in itchiness, spots and pus blisters around sweat glands. Workers exposed to antimony can develop obstructive lung disease. Antimony crosses the placenta and is excreted in breast milk. There may be an increased incidence of spontaneous late abortions, premature births, and gynaecological problems among female antimony smelter workers. Antimony may be associated with an increased incidence of cancers of the lung and chest.

### **TOXICITY AND IRRITATION**

■ The substance is classified by IARC as Group 3:

NOT classifiable as to its carcinogenicity to humans.

Evidence of carcinogenicity may be inadequate or limited in animal testing.

#### **CARCINOGEN**

antimony trisulfide

International Agency for Research on Cancer (IARC) -Agents Reviewed by the IARC Monographs

Group

3

Not classifiable as to its carcinogenicity to humans

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### **Section 12 - ECOLOGICAL INFORMATION**

Very toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment.

This material and its container must be disposed of as hazardous waste.

Avoid release to the environment.

Refer to special instructions/ safety data sheets.

**Ecotoxicity** 

Ingredient Persistence: Persistence: Air Bioaccumulation Mobility
Water/Soil
antimony trisulfide No Data No Data No Data Available Available Available Available

### **Section 13 - DISPOSAL CONSIDERATIONS**

- Containers may still present a chemical hazard/ danger when empty.
- Return to supplier for reuse/ recycling if possible.

Otherwise:

- If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill.
- Where possible retain label warnings and MSDS and observe all notices pertaining to the product.

Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area.

A Hierarchy of Controls seems to be common - the user should investigate:

- · Reduction.
- DO NOT allow wash water from cleaning or process equipment to enter drains.
- It may be necessary to collect all wash water for treatment before disposal.
- In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.
- Where in doubt contact the responsible authority.
- Recycle wherever possible or consult manufacturer for recycling options.
- Consult State Land Waste Management Authority for disposal.
- Bury residue in an authorised landfill.
- Recycle containers if possible, or dispose of in an authorised landfill.

### Section 14 - TRANSPORTATION INFORMATION





Labels Required: TOXIC

### HAZCHEM:

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Land Transport UNDG:

Class or division: 6.1 Subsidiary risk: None UN No.: 1549 UN packing group: III

Shipping Name: ANTIMONY COMPOUND, INORGANIC, SOLID, N.O.S. (contains

antimony trisulfide)

Air Transport IATA:

ICAO/IATA Class: 6.1 ICAO/IATA Subrisk: None UN/ID Number: 1549 Packing Group: III Special provisions: A12

Shipping name: ANTIMONY COMPOUND, INORGANIC, SOLID, N.O.S. (contains antimony trisulfide)

**Maritime Transport IMDG:** 

IMDG Class:6.1IMDG Subrisk:NoneUN Number:1549Packing Group:IIIEMS Number:F- A, S- ASpecial provisions:45 274Limited Quantities:5 kgMarine Pollutant:Yes

Shipping name: ANTIMONY COMPOUND, INORGANIC, SOLID, N.O.S. (contains antimony trisulfide)

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Section 14 - TRANSPORTATION INFORMATION

A12 (45) Antimony sulphides and oxides which contain 0.5% or less of arsenic calculated on the total weight are not subject to these Regulations.

# **Section 15 - REGULATORY INFORMATION**

### **Section 16 - OTHER INFORMATION**

■ The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings.

The above information is believed to be accurate and represent the best information currently available to us, but does not represent any warranty expressed or implied of the properties of the product. User should make their own investigation to determine the suitability of the information for their particular purpose.

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